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Description: Biochemistry: The Molecular Basis of Life is a one-semester text focusing on the essential biochemical principles that underpin the modern life sciences. The sixth edition offers deeper coverage of the chemistry of reactions while emphasizing the relationship between biochemistry and human biology.

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Biochemistry is a broad discipline that is growing rapidly in its scope - new developments and discoveries are being made everyday. The goal of this class will be to give you a solid background with which you can appreciate the latest developments and research papers.

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Note that the term molecular mass is preferred to the older term molecular weight. The SI unit of concentration is expressed in terms of moles per cubic metre ( $\text{mol m}^{-3}$ ) (see Table 1.1). In practice this is far too large for normal laboratory purposes and a unit based on a cubic decimetre ( $\text{dm}^3, 10^{-3} \text{m}^3$ ) is preferred.

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The inner part of glucose units in amylopectin are joined by  $\alpha$ -1,4 glycosidic linkage as in amylose, but the branch points of amylopectin are  $\alpha$ -1,6 linkages. The branch points repeat about every 20 to 30 (1-4) linkages. Glycogen. - Glycogen is the main storage polysaccharide of animal cells (Animal starch).

### **lecnote fm Med Biochem - The Carter Center**

Biochemistry can be defined as the science concerned with the chemical basis of life. The cell is the structural unit of living systems. Thus, biochemistry can also be described as the science concerned with the chemical constituents of living cells and with reactions and processes they undergo.

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